# **ChemWatch Review SDS**

Chemwatch: 1013 Version No: 9.1

Safety Data Sheet according to WHS Regulations (Hazardous Chemicals) Amendment 2020 and ADG requirements

# Chemwatch Hazard Alert Code: 1

Issue Date: 20/12/2017 Print Date: 16/05/2022 S.GHS.AUS.EN

# SECTION 1 Identification of the substance / mixture and of the company / undertaking

# Product Identifier

Product name	CARBON DIOXIDE
Chemical Name	carbon dioxide
Synonyms	CO2; carbonic anhydride; carbonic acid gas; carbon dioxide propellant; carbon dioxide liquid; carbonoxide (CO2); carbon-12 dioxide; carbon-12C dioxide-16O2; carbonic acid anhydride; dry ice; EN 439C1; Khladon744; R 744; 080 Welding Grade 081 Industrial Grade 082 Food Grade 083 Hotel Grade; 084 Beergas; 086 McDonald's Grade; 176 Anaerobic Grade; 200 Fogg - Pesticide Grade; Grades 131 206 207 261; Aligal; lasal; Refrigerant 744; BOC Gases Pack Sizes:; Gas withdrawal; 6 kg (D), 10 kg (VT), 15 kg (E), 22 kg (F), 31 kg (G); Liquid withdrawal: 6 kg (DE), 15 kg (EE), 31 kg (GE); carbon dioxide, food grade; carbon dioxide, wine grade; carbon dioxide bulk; YY-RES-34452; carbon dioxide, compressed; Carbon Dioxide Cylinder, Gas and Liquid Withdrawal. Product code: 080, 081, 082, 083, 084, 086, 131, 176, 200. Refrigerant 744.; carbon dioxide, compressed (CO2)
Proper shipping name	CARBON DIOXIDE
Chemical formula	CO2
Other means of identification	Not Available
CAS number	124-38-9

# Relevant identified uses of the substance or mixture and uses advised against

Relevant identified uses

Used in the carbonation of beverages; welding shielding gas; inert atmosphere; manufacture of carbonates; in fire prevention and fire extinction; a rice fumigant; as an antiseptic in bacteriology and in the frozen food industry; refrigerant.

# Details of the supplier of the safety data sheet

Registered company name	BOC (a division of Linde)	Air Liquide	Thales Australia, Benalla	
Address	10 Julius Avenue North Ryde NSW 2113 Australia	3 Channel Close Henderson WA 6166 Australia	PO Box 840, Benalla-Yarrawonga Rd Benalla VIC 2671 Australia	
Telephone	+61 2 8874 4400	+61 8 9494 9600	+61 3 5760 3222 +61 3 5760 3553	
Fax	+61 2 9886 9000	+61 8 9494 9682	+61 3 5760 3233	
Website	http://pgw100.portal.gases.boc.com /scripts/wgate/zcpwp_b2c/!?~	http://industry.airliquide.com.au	Not Available	
Email	Chris.Sherry@boc.com	ALWACST@airliquide.com	support.internet@thalesgroup.com	

Registered company name	Ultra Electronics (Ultra Electronics - Avalon Systems)	Leland
Address	12 Douglas Drive, Technology Park Mawson Lakes SA 5095 Australia	2614 South Clinton Avenue South Plainfield NJ 07080 United States
Telephone	+61 8 8169 1200	+1 908 668 1008 +1 800 984 9793
Fax	+61 8 8169 1201	+1 908 668 7716
Website	www.ultra-electronics.com	http://www.lelandltd.com/
Email	info@ultra-avalon.com	tech@lelandltd.com

# **Emergency telephone number**

Association / Organisation	BOC (a division of Linde) Air Liquide Poison I		Poison Information Centre (Australia)	
Emergency telephone numbers	1800 653 572 (A/H) (Australia only)	1800 812 588	13 11 26 (24 hours)	
Other emergency telephone numbers	Not Available	Not Available	Not Available	
Association / Organisation	CHEMWATCH EMERGENCY RESPONSE			

Issue Date: **20/12/2017**Print Date: **16/05/2022** 

Emergency telephone numbers	+61 1800 951 288
Other emergency telephone numbers	+61 3 9573 3188

Once connected and if the message is not in your prefered language then please dial 01

### **SECTION 2 Hazards identification**

### Classification of the substance or mixture

Poisons Schedule	Not Applicable
Classification [1]	Gases Under Pressure (Liquefied Gas)
Legend:	1. Classified by Chemwatch; 2. Classification drawn from HCIS; 3. Classification drawn from Regulation (EU) No 1272/2008 - Annex VI

### Label elements

Hazard pictogram(s)



Signal word

Warning

# Hazard statement(s)

H280

Contains gas under pressure; may explode if heated.

# Precautionary statement(s) Prevention

Not Applicable

# Precautionary statement(s) Response

Not Applicable

### Precautionary statement(s) Storage

P410+P403

Protect from sunlight. Store in a well-ventilated place.

# Precautionary statement(s) Disposal

Not Applicable

Not Applicable

### **SECTION 3 Composition / information on ingredients**

### **Substances**

CAS No	%[weight]	Name
124-38-9	>99.8	carbon dioxide

Legend:

1. Classified by Chemwatch; 2. Classification drawn from HCIS; 3. Classification drawn from Regulation (EU) No 1272/2008 - Annex VI; 4. Classification drawn from C&L; \* EU IOELVs available

### **Mixtures**

See section above for composition of Substances

### **SECTION 4 First aid measures**

### Description of first aid measures

For exposure to dry ice powder: Keep patient calm. Immediately flush eye with tepid water in large quantities, or with sterile saline solution.

# **Eye Contact**

If product comes in contact with eyes remove the patient from gas source or contaminated area.

► Take the patient to the nearest eye wash, shower or other source of clean water.

 Chemwatch: 1013
 Page 3 of 11
 Issue Date: 20/12/2017

 Version No: 9.1
 CARBON DIOXIDE
 Print Date: 16/05/2022

	<ul> <li>Open the eyelid(s) wide to allow the material to evaporate.</li> <li>Gently rinse the affected eye(s) with clean, cool water for at least 15 minutes. Have the patient lie or sit down and tilt the head back. Hold the eyelid(s) open and pour water slowly over the eyeball(s) at the inner corners, letting the water run out of the outer corners.</li> <li>The patient may be in great pain and wish to keep the eyes closed. It is important that the material is rinsed from the eyes to prevent further damage.</li> <li>Ensure that the patient looks up, and side to side as the eye is rinsed in order to better reach all parts of the eye(s)</li> <li>Transport to hospital or doctor.</li> <li>Even when no pain persists and vision is good, a doctor should examine the eye as delayed damage may occur.</li> <li>If the patient cannot tolerate light, protect the eyes with a clean, loosely tied bandage.</li> <li>Ensure verbal communication and physical contact with the patient.</li> <li>DO NOT allow the patient to rub the eyes</li> <li>DO NOT allow the patient to tightly shut the eyes</li> <li>DO NOT introduce oil or ointment into the eye(s) without medical advice</li> <li>DO NOT use hot or tepid water.</li> </ul>
Skin Contact	If skin or hair contact occurs:  Flush skin and hair with running water (and soap if available).  Seek medical attention in event of irritation.  In case of cold burns (frost-bite):  Move casualty into warmth before thawing the affected part; if feet are affected carry if possible  Bathe the affected area immediately in luke-warm water (not more than 35 deg C) for 10 to 15 minutes, immersing if possible and without rubbing  DO NOT apply hot water or radiant heat.  Apply a clean, dry, light dressing of "fluffed-up" dry gauze bandage  If a limb is involved, raise and support this to reduce swelling  If an adult is involved and where intense pain occurs provide pain killers such as paracetomol  Transport to hospital, or doctor  Subsequent blackening of the exposed tissue indicates potential of necrosis, which may require amputation.
Inhalation	<ul> <li>Following exposure to gas, remove the patient from the gas source or contaminated area.</li> <li>NOTE: Personal Protective Equipment (PPE), including positive pressure self-contained breathing apparatus may be required to assure the safety of the rescuer.</li> <li>Prostheses such as false teeth, which may block the airway, should be removed, where possible, prior to initiating first aid procedures.</li> <li>If the patient is not breathing spontaneously, administer rescue breathing.</li> <li>If the patient does not have a pulse, administer CPR.</li> <li>If medical oxygen and appropriately trained personnel are available, administer 100% oxygen.</li> <li>Summon an emergency ambulance. If an ambulance is not available, contact a physician, hospital, or Poison Control Centre for further instruction.</li> <li>Keep the patient warm, comfortable and at rest while awaiting medical care.</li> <li>MONITOR THE BREATHING AND PULSE, CONTINUOUSLY.</li> <li>Administer rescue breathing (preferably with a demand-valve resuscitator, bag-valve mask-device, or pocket mask as trained) or CPR if necessary.</li> </ul>
Ingestion	Not considered a normal route of entry.

### Indication of any immediate medical attention and special treatment needed

Persons with cardiovascular or pulmonary disease are at increased risk. Prolonged breathing of cold vapour may damage lung tissue.

For frost-bite caused by liquefied petroleum gas:

- If part has not thawed, place in warm water bath (41-46 C) for 15-20 minutes, until the skin turns pink or red.
- Analgesia may be necessary while thawing.
- If there has been a massive exposure, the general body temperature must be depressed, and the patient must be immediately rewarmed by whole-body immersion, in a bath at the above temperature.
- Shock may occur during rewarming.
- Administer tetanus toxoid booster after hospitalization.
- Prophylactic antibiotics may be useful.
- ▶ The patient may require anticoagulants and oxygen.

[Shell Australia 22/12/87]

For gas exposures:

# BASIC TREATMENT

- Establish a patent airway with suction where necessary.
- Watch for signs of respiratory insufficiency and assist ventilation as necessary.
- Administer oxygen by non-rebreather mask at 10 to 15 l/min.
- $\mbox{\ensuremath{\,^{\blacktriangleright}}}$  Monitor and treat, where necessary, for pulmonary oedema .
- ▶ Monitor and treat, where necessary, for shock.
- ► Anticipate seizures.

ADVANCED TREATMENT

Chemwatch: 1013 Version No: 9.1

Page 4 of 11 Issue Date: 20/12/2017 Print Date: 16/05/2022 **CARBON DIOXIDE** 

- b Consider orotracheal or nasotracheal intubation for airway control in unconscious patient or where respiratory arrest has occurred.
- Positive-pressure ventilation using a bag-valve mask might be of use.
- Monitor and treat, where necessary, for arrhythmias.
- Start an IV D5W TKO. If signs of hypovolaemia are present use lactated Ringers solution. Fluid overload might create complications.
- Drug therapy should be considered for pulmonary oedema.
- Hypotension with signs of hypovolaemia requires the cautious administration of fluids. Fluid overload might create complications.
- ► Treat seizures with diazepam.
- ▶ Proparacaine hydrochloride should be used to assist eye irrigation.

BRONSTEIN, A.C. and CURRANCE, P.L.

EMERGENCY CARE FOR HAZARDOUS MATERIALS EXPOSURE: 2nd Ed. 1994

# **SECTION 5 Firefighting measures**

### **Extinguishing media**

SMALL FIRE: Use extinguishing agent suitable for type of surrounding fire.

LARGE FIRE: Cool cylinder.

DO NOT direct water at source of leak or venting safety devices as icing may occur.

### Special hazards arising from the substrate or mixture

Fire Incompatibility	None known.
Advice for firefighters	
Fire Fighting	GENERAL  Alert Fire Brigade and tell them location and nature of hazard.  Wear breathing apparatus and protective gloves.  Fight fire from a safe distance, with adequate cover.  Use water delivered as a fine spray to control fire and cool adjacent area.
Fire/Explosion Hazard	Non combustible. Not considered a significant fire risk, however containers may burn.

Vented gas is more dense than air and may collect in pits, basements.

Contains low boiling substance: Closed containers may rupture due to pressure buildup under fire conditions.

### **SECTION 6 Accidental release measures**

**HAZCHEM** 

## Personal precautions, protective equipment and emergency procedures

See section 8

# **Environmental precautions**

See section 12

### Methods and material for containment and cleaning up

2T

Minor Spills	<ul> <li>Avoid breathing vapour and any contact with liquid or gas. Protective equipment including respirator should be used.</li> <li>DO NOT enter confined spaces where gas may have accumulated.</li> <li>Increase ventilation.</li> </ul>
Major Spills	<ul> <li>Clear area of all unprotected personnel and move upwind.</li> <li>Alert Emergency Authority and advise them of the location and nature of hazard.</li> <li>Wear breathing apparatus and protective gloves.</li> <li>Prevent by any means available, spillage from entering drains and water-courses.</li> <li>Remove leaking cylinders to a safe place.</li> <li>Fit vent pipes. Release pressure under safe, controlled conditions</li> <li>Burn issuing gas at vent pipes.</li> <li>DO NOT exert excessive pressure on valve; DO NOT attempt to operate damaged valve.</li> </ul>

Personal Protective Equipment advice is contained in Section 8 of the SDS.

### **SECTION 7 Handling and storage**

# Precautions for safe handling

Consider use in closed pressurised systems, fitted with temperature, pressure and safety relief valves which are vented for Safe handling

Issue Date: **20/12/2017**Print Date: **16/05/2022** 

safe dispersal. Use only properly specified equipment which is suitable for this product, its supply pressure and temperature

- · The tubing network design connecting gas cylinders to the delivery system should include appropriate pressure indicators and vacuum or suction lines.
- · Fully-welded types of pressure gauges, where the bourdon tube sensing element is welded to the gauge body, are recommended.
- Before connecting gas cylinders, ensure manifold is mechanically secure and does not containing another gas.
- ▶ DO NOT transfer gas from one cylinder to another.
- Store in an upright position.
  - Cylinders should be stored in a purpose-built compound with good ventilation, preferably in the open.
    - ▶ Such compounds should be sited and built in accordance with statutory requirements.
    - ▶ The storage compound should be kept clear and access restricted to authorised personnel only.
    - ▶ Cylinders stored in the open should be protected against rust and extremes of weather.

### Conditions for safe storage, including any incompatibilities

### Suitable container

Other information

- Cylinder:
- Ensure the use of equipment rated for cylinder pressure.
- Ensure the use of compatible materials of construction.
- Valve protection cap to be in place until cylinder is secured, connected.
- ▶ Cylinder must be properly secured either in use or in storage.

# Carbon dioxide:

- reacts violently with strong bases and alkali metals (especially their dusts)
- may ignite or explode when heated or in suspended chemically active metals (and their hydrides) such as aluminium, chromium, manganese, magnesium (above 775 C), titanium (above 550 C), uranium (above 750 C) or zirconium, diethylmagnesium

### Storage incompatibility

- is incompatible with water, acrolein, acrylaldehyde, amines, anhydrous ammonia, aziridine, metal acetylides (such as lithium acetylide), caesium monoxide (moist), lithium, potassium, sodium, sodium carbide, sodium-potassium alloy, sodium peroxide, titanium
- may build up static electricity when discharged at high flow rates from storage cylinders or fire extinguishers this may produce sparks resulting in ignition of flammables or explosives.
- may decompose to toxic carbon monoxide and flammable oxygen when exposed to electrical discharges or very high temperatures
- Compressed gases may contain a large amount of kinetic energy over and above that potentially available from the energy of reaction produced by the gas in chemical reaction with other substances

### **SECTION 8 Exposure controls / personal protection**

### **Control parameters**

### Occupational Exposure Limits (OEL)

### **INGREDIENT DATA**

Source	Ingredient	Material name	TWA	STEL	Peak	Notes
Australia Exposure	carbon	Carbon dioxide in coal mines	12500 ppm / 22500	54000 mg/m3 / 30000	Not	Not
Standards	dioxide		mg/m3	ppm	Available	Available
Australia Exposure	carbon	Carbon dioxide	5000 ppm / 9000	54000 mg/m3 / 30000	Not	Not
Standards	dioxide		mg/m3	ppm	Available	Available

# **Emergency Limits**

Ingredient	TEEL-1	TEEL-2	TEEL-3
CARBON DIOXIDE	Not Available	Not Available	Not Available

Ingredient	Original IDLH	Revised IDLH
carbon dioxide	40,000 ppm	Not Available

## **Exposure controls**

# Appropriate engineering controls

Engineering controls are used to remove a hazard or place a barrier between the worker and the hazard. Well-designed engineering controls can be highly effective in protecting workers and will typically be independent of worker interactions to provide this high level of protection.

The basic types of engineering controls are:

Process controls which involve changing the way a job activity or process is done to reduce the risk.

Enclosure and/or isolation of emission source which keeps a selected hazard "physically" away from the worker and ventilation that strategically "adds" and "removes" air in the work environment.

Issue Date: **20/12/2017**Print Date: **16/05/2022** 

Personal protection Chemical goggles. Full face shield may be required for supplementary but never for primary protection of eyes. Eye and face protection ▶ Contact lenses may pose a special hazard; soft contact lenses may absorb and concentrate irritants. A written policy document, describing the wearing of lenses or restrictions on use, should be created for each workplace or task. Skin protection See Hand protection below ▶ When handling sealed and suitably insulated cylinders wear cloth or leather gloves. Insulated gloves: Hands/feet protection NOTE: Insulated gloves should be loose fitting so that may be removed quickly if liquid is spilled upon them. Insulated gloves are not made to permit hands to be placed in the liquid; they provide only short-term protection from accidental contact with the liquid. **Body protection** See Other protection below ▶ Protective overalls, closely fitted at neck and wrist. ► Eye-wash unit. Other protection ▶ Ensure availability of lifeline in confined spaces. Staff should be trained in all aspects of rescue work.

### Respiratory protection

Full face respirator with supplied air.

Positive pressure, full face, air-supplied breathing apparatus should be used for work in enclosed spaces if a leak is suspected or the primary containment is to be opened (e.g. for a cylinder change)

A colourless, odourless gas; slightly soluble in water. A saturated solution of gas in water has pH of 3.8. High pressure liquefiable

VOC g/L

Not Available

Air-supplied breathing apparatus is required where release of gas from primary containment is either suspected or demonstrated.

das: Critical temperature 31 ded. C approx

# **SECTION 9 Physical and chemical properties**

**Appearance** 

### Information on basic physical and chemical properties

	gas: Critical temperature 31 deg. C approx.		
Physical state	Liquified Gas	Relative density (Water = 1)	1.10 @ -37 C
Odour	Not Available	Partition coefficient n-octanol / water	Not Available
Odour threshold	Not Available	Auto-ignition temperature (°C)	Not Applicable
pH (as supplied)	Not Applicable	Decomposition temperature	Not Applicable
Melting point / freezing point (°C)	-78.5	Viscosity (cSt)	0.015(101 kPa,26 C)
Initial boiling point and boiling range (°C)	-56.6 (526.9 kPa)	Molecular weight (g/mol)	44
Flash point (°C)	Not Applicable	Taste	Not Available
Evaporation rate	Not Available	Explosive properties	Not Available
Flammability	Not Applicable	Oxidising properties	Not Available
Upper Explosive Limit (%)	Not Applicable	Surface Tension (dyn/cm or mN/m)	Not Available
Lower Explosive Limit (%)	Not Applicable	Volatile Component (%vol)	100
Vapour pressure (kPa)	6300 @ 25 C	Gas group	Not Available
Solubility in water	Partly miscible	pH as a solution (Not Available%)	Not Available

### **SECTION 10 Stability and reactivity**

Vapour density (Air = 1)

Reactivity	See section 7
------------	---------------

Issue Date: **20/12/2017**Print Date: **16/05/2022** 

Chemical stability	<ul> <li>Unstable in the presence of incompatible materials.</li> <li>Product is considered stable.</li> <li>Hazardous polymerisation will not occur.</li> </ul>
Possibility of hazardous reactions	See section 7
Conditions to avoid	See section 7
Incompatible materials	See section 7
Hazardous decomposition products	See section 5

### **SECTION 11 Toxicological information**

Inhaled

Information on tox	icological	effects
--------------------	------------	---------

The material is not thought to produce respiratory irritation (as classified by EC Directives using animal models). Nevertheless
inhalation of the material, especially for prolonged periods, may produce respiratory discomfort and occasionally, distress.
Inhalation of vapours may cause drowsiness and dizziness. This may be accompanied by sleepiness, reduced alertness, loss of
reflexes, lack of co-ordination, and vertigo.
Inhalation of vapours or aerosols (mists, fumes), generated by the material during the course of normal handling, may be
damaging to the health of the individual.
Carbon dioxide is an odourless gas, which gives very poor warning of exposure. It can cause rapid loss of consciousness, and
death from lack of oxygen at concentrations of 10% in air.
Carbon dioxide is the most powerful dilator of brain vessels known.
Material is highly volatile and may quickly form a concentrated atmosphere in confined or unventilated areas. The vapour may

Material is highly volatile and may quickly form a concentrated atmosphere in confined or unventilated areas. The vapour may displace and replace air in breathing zone, acting as a simple asphyxiant. This may happen with little warning of overexposure. Symptoms of asphyxia (suffocation) may include headache, dizziness, shortness of breath, muscular weakness, drowsiness and ringing in the ears. If the asphyxia is allowed to progress, there may be nausea and vomiting, further physical weakness and unconsciousness and, finally, convulsions, coma and death.

	unconsciousness and, finally, convulsions, coma and death.
Ingestion	Solid carbon dioxide will cause severe cold burns to mouth and throat. Overexposure is unlikely in this form.  Not normally a hazard due to physical form of product.
	Considered an unlikely route of entry in commercial/industrial environments

The material is not thought to produce adverse health effects or skin irritation following contact (as classified by EC Directives using animal models). Nevertheless, good hygiene practice requires that exposure be kept to a minimum and that suitable gloves be used in an occupational setting.

Open cuts, abraded or irritated skin should not be exposed to this material

# Skin Contact

Entry into the blood-stream, through, for example, cuts, abrasions or lesions, may produce systemic injury with harmful effects. Examine the skin prior to the use of the material and ensure that any external damage is suitably protected.

Vapourising liquid causes rapid cooling and contact may cause cold burns, frostbite, even through normal gloves. Frozen skin tissues are painless and appear waxy and yellow. Signs and symptoms of frost-bite may include "pins and needles", paleness followed by numbness, a hardening an stiffening of the skin, a progression of colour changes in the affected area, (first white, then mottled and blue and eventually black; on recovery, red, hot, painful and blistered).

### Eye

Although the material is not thought to be an irritant (as classified by EC Directives), direct contact with the eye may produce transient discomfort characterised by tearing or conjunctival redness (as with windburn).

Substance accumulation, in the human body, may occur and may cause some concern following repeated or long-term occupational exposure.

### Chronic

There is some evidence from animal testing that exposure to this material may result in reduced fertility.

Not considered to be a risk because of the extreme volatility of the gas.

There is some evidence from animal testing that exposure to this material may result in toxic effects to the unborn baby. Main route of exposure to the gas in the workplace is by inhalation.

Although long-term exposure to carbon dioxide, at levels up to 1.5% in inhaled air, are well tolerated, the metabolism of calcium/phosphorus metabolism may be affected. Calcium levels in the blood and phosphorus in the urine progressively fall.

	TOXICITY	IRRITATION	
carbon dioxide	Not Available	Not Available	
Legend:	Value obtained from Europe ECHA Registered Substances - Acute toxicity 2.* Value obtained from manufacturer's SDS.     Unless otherwise specified data extracted from RTECS - Register of Toxic Effect of chemical Substances		

Acute Toxicity	×	Carcinogenicity	×
Skin Irritation/Corrosion	×	Reproductivity	×
Serious Eye Damage/Irritation	×	STOT - Single Exposure	×
Respiratory or Skin sensitisation	×	STOT - Repeated Exposure	×

Issue Date: 20/12/2017 Print Date: 16/05/2022

### **CARBON DIOXIDE**

Mutagenicity

**Aspiration Hazard** 

★ - Data either not available or does not fill the criteria for classification

✓ – Data available to make classification

# **SECTION 12 Ecological information**

### **Toxicity**

aankan diawida	Endpoint	Test Duration (hr)	Species	Value	Source
carbon dioxide	LC50	96h	Fish	35mg/l	1
Legend:	Extracted from 1. IUCLID Toxicity Data 2. Europe ECHA Registered Substances - Ecotoxicological Information - Aquatic Toxicity 4. US EPA, Ecotox database - Aquatic Toxicity Data 5. ECETOC Aquatic Hazard Assessment Data 6. NITE (Japan) - Bioconcentration Data 7. METI (Japan) - Bioconcentration Data 8. Vendor Data				

Legend:

#### For carbon dioxide:

Environmental Fate: Carbon dioxide in earth's atmosphere is considered a trace gas. There are seasonal fluctuations of atmospheric concentrations of carbon dioxide primarily due to CO2 absorbed during seasonal plant growth. Due to human activities such as the combustion of fossil fuels and deforestation, the concentration of atmospheric carbon dioxide has increased by about 35% since preindustrial times. Carbon dissolved in the oceans is about 50 times greater than CO2 found in the atmosphere.

# Persistence and degradability

Ingredient	Persistence: Water/Soil	Persistence: Air
carbon dioxide	LOW	LOW

### **Bioaccumulative potential**

Ingredient	Bioaccumulation
carbon dioxide	LOW (LogKOW = 0.83)

# Mobility in soil

Ingredient	Mobility
carbon dioxide	HIGH (KOC = 1.498)

# **SECTION 13 Disposal considerations**

## Waste treatment methods

**Product / Packaging** disposal

- Evaporate residue at an approved site.
- PReturn empty containers to supplier. If containers are marked non-returnable establish means of disposal with manufacturer prior to purchase.
- ► Ensure damaged or non-returnable cylinders are gas-free before disposal.

# **SECTION 14 Transport information**

### **Labels Required**



# Land transport (ADG)

UN number	1013		
UN proper shipping name	CARBON DIOXIDE		
Transport hazard class(es)	Class	2.2	
	Subrisk	Not Applicable	

Issue Date: **20/12/2017**Print Date: **16/05/2022** 

Packing group	Not Applicable		
Environmental hazard	Not Applicable		
Special precautions for user	· · ·	378 392 120 ml	

## Air transport (ICAO-IATA / DGR)

UN number	1013		
UN proper shipping name	Carbon dioxide		
Transport hazard class(es)	ICAO/IATA Class	2.2	
	ICAO / IATA Subrisk	Not Applicable	
	ERG Code 2L		
Packing group	Not Applicable		
Environmental hazard	Not Applicable		
	Special provisions	A202	
	Cargo Only Packing Instructions		200
Special precautions for user	Cargo Only Maximum Qty / Pack		150 kg
	Passenger and Cargo Packing Instructions		200
	Passenger and Cargo Maximum Qty / Pack		75 kg
	Passenger and Cargo Limited Quantity Packing Instructions		Forbidden
	Passenger and Cargo Limited Maximum Qty / Pack		Forbidden

# Sea transport (IMDG-Code / GGVSee)

UN number	1013			
UN proper shipping name	CARBON DIOXIDE	CARBON DIOXIDE		
Transport hazard class(es)		2.2 Not Applicable		
Packing group	Not Applicable			
Environmental hazard	Not Applicable			
Special precautions for user	EMS Number Special provisions Limited Quantities			

# Transport in bulk according to Annex II of MARPOL and the IBC code

Not Applicable

# Transport in bulk in accordance with MARPOL Annex V and the IMSBC Code

Product name	Group
carbon dioxide	Not Available

# Transport in bulk in accordance with the ICG Code

Product name	Ship Type
carbon dioxide	Not Available

# **SECTION 15 Regulatory information**

# Safety, health and environmental regulations / legislation specific for the substance or mixture

carbon dioxide is found on the following regulatory lists

Australian Inventory of Industrial Chemicals (AIIC)

FEI Equine Prohibited Substances List (EPSL)

Issue Date: **20/12/2017**Print Date: **16/05/2022** 

# **National Inventory Status**

National Inventory	Status
Australia - AIIC / Australia Non-Industrial Use	Yes
Canada - DSL	Yes
Canada - NDSL	No (carbon dioxide)
China - IECSC	Yes
Europe - EINEC / ELINCS / NLP	Yes
Japan - ENCS	Yes
Korea - KECI	Yes
New Zealand - NZIoC	Yes
Philippines - PICCS	Yes
USA - TSCA	Yes
Taiwan - TCSI	Yes
Mexico - INSQ	Yes
Vietnam - NCI	Yes
Russia - FBEPH	Yes
Legend:	Yes = All CAS declared ingredients are on the inventory No = One or more of the CAS listed ingredients are not on the inventory. These ingredients may be exempt or will require registration.

### **SECTION 16 Other information**

Revision Date	20/12/2017
Initial Date	14/06/2005

### **SDS Version Summary**

Version	Date of Update	Sections Updated
8.1	23/03/2015	Acute Health (skin), Acute Health (swallowed), Advice to Doctor, First Aid (skin), Personal Protection (eye), Personal Protection (hands/feet), Physical Properties, Storage (storage incompatibility), Supplier Information
9.1	20/12/2017	Toxicity and Irritation (Other)

### Other information

Classification of the preparation and its individual components has drawn on official and authoritative sources as well as independent review by the Chemwatch Classification committee using available literature references.

The SDS is a Hazard Communication tool and should be used to assist in the Risk Assessment. Many factors determine whether the reported Hazards are Risks in the workplace or other settings. Risks may be determined by reference to Exposures Scenarios. Scale of use, frequency of use and current or available engineering controls must be considered.

# **Definitions and abbreviations**

PC-TWA: Permissible Concentration-Time Weighted Average

PC-STEL: Permissible Concentration-Short Term Exposure Limit

IARC: International Agency for Research on Cancer

ACGIH: American Conference of Governmental Industrial Hygienists

STEL: Short Term Exposure Limit

TEEL: Temporary Emergency Exposure Limit。

IDLH: Immediately Dangerous to Life or Health Concentrations

ES: Exposure Standard OSF: Odour Safety Factor

NOAEL :No Observed Adverse Effect Level LOAEL: Lowest Observed Adverse Effect Level

TLV: Threshold Limit Value LOD: Limit Of Detection OTV: Odour Threshold Value BCF: BioConcentration Factors Chemwatch: 1013 Page 11 of 11 Issue Date: 20/12/2017 Version No: 9.1 Print Date: 16/05/2022

### **CARBON DIOXIDE**

BEI: Biological Exposure Index

AIIC: Australian Inventory of Industrial Chemicals

DSL: Domestic Substances List NDSL: Non-Domestic Substances List

IECSC: Inventory of Existing Chemical Substance in China

EINECS: European INventory of Existing Commercial chemical Substances

ELINCS: European List of Notified Chemical Substances

NLP: No-Longer Polymers

**ENCS: Existing and New Chemical Substances Inventory** 

KECI: Korea Existing Chemicals Inventory NZIoC: New Zealand Inventory of Chemicals

PICCS: Philippine Inventory of Chemicals and Chemical Substances

TSCA: Toxic Substances Control Act

TCSI: Taiwan Chemical Substance Inventory INSQ: Inventario Nacional de Sustancias Químicas

NCI: National Chemical Inventory

FBEPH: Russian Register of Potentially Hazardous Chemical and Biological Substances

### This document is copyright.

Apart from any fair dealing for the purposes of private study, research, review or criticism, as permitted under the Copyright Act, no part may be reproduced by any process without written permission from CHEMWATCH.

TEL (+61 3) 9572 4700.